CHM 113 INA Laboratory Discussion
FALL 2015

The Discussion (online) component is part of the CHM 113 Laboratory.
Online Course: Check course website and LIVE-D2L each week

Instructor: Dr. Adriana Dinescu
Office: Room CSC 408
Off. Hrs: M R 3:00 PM - 5:00 PM
R 9:00 AM - 10:00 AM, or by appointment
Office Ph.: 408-4623
E-mail: adriana.dinescu@wilkes.edu

Course website: www.chem.wilkes.edu → follow the Resources tab, select Links and click on CHM 113 link.
The URL is: http://www.chem.wilkes.edu/~dinescu/CHM113

Attendance: Online attendance is mandatory and will be monitored in LIVE D2L. While this course is delivered online, you are required to learn each module (one module for each experiment) before performing the laboratory experiment.

Instructions and announcements will be posted on the course website: http://www.chem.wilkes.edu/~dinescu/CHM113

Schedule: Laboratory schedule is printed in the lab manual and posted on the course website. Any changes will be announced and online schedule will reflect those changes.

Course materials will be delivered on Wednesday, the week before the scheduled experiment. For example, if your “Measurements” lab is scheduled on Tuesday, September 8, the course materials for this experiment will be posted online on Wednesday, September 2nd. This way you will have five days to prepare.

Assessment: There are no exams or grades for the Discussion. However, your performance in the laboratory will be evaluated in connection with your progress in the online lab discussion. The lab grade has one component that accounts for 12.5% of your final grade (attendance, technique, performance and professionalism). This component will take into account your online attendance and level of preparation for the scheduled experiment. Your lab instructor will know the content of the materials that were delivered in the online course and expect that you are familiar with all of them.

Lab requirements: Follow the laboratory syllabus given by your lab instructor.

Prelab questions are due before the scheduled lab. Questions are provided in the lab manual (end of the experiment).
Before the scheduled lab, you should write in your notebook the title, purpose, materials and procedure for that experiment. It saves time and frustration to prepare your experiment by studying the lab manual and online materials before coming to lab. You will not be allowed to use the manual while performing the lab.

Lab report forms will be posted during the week of scheduled lab. Do not copy the template of the report form in your lab notebook. You need to record the raw data based on actual procedure and not the expected/planned procedure.

Office hours policies:
- If you have questions that would require some discussions, I will not be able to answer through email; please follow the office hours listed on the first page of the syllabus.
- My office hours are meant for you; use them as much as you need to.
- If you have classes during my office hours, please schedule an appointment. Do not ask when I am available. Propose a day/time that works with your schedule and mine (my schedule is printed on page 3). As you can see from my schedule, the best options are weekdays after 5 pm.
- If you have a question that is related to the online component and can be answered in 1-3 sentences (such as a short clarification), please feel free to email me and I will try to provide an answer within 24 hours.

Major Objectives (shared by all courses in the Department of Chemistry):
A1. To demonstrate proficiency in analysis, organization, interpretation, and presentation of chemical data
A2. To express chemical concepts with quantitative relationships and to interpret the results obtained from the use of these quantitative relationships in terms of the chemical concepts conveyed in this format
A3. To use written communication in a cogent and coherent form that demonstrates understanding of chemical concepts
A4. To develop critical thinking skills in synthesizing information
A5. To appreciate the relevance of chemistry to everyday life
A6. To recognize that the various areas of chemistry are interrelated and require integration of basic chemical principles including chemical formulas and nomenclature; chemical reactions and stoichiometry; chemical equilibria including acid-base theory; and molecular structure

Course Objectives (for this particular course):
B1. To reinforce or extend the chemical theories discussed in the co-requisite and all prerequisite chemistry courses
B2. To develop proficiency in experimental design, data collection, and analysis and interpretation of experimental results
B3. To become familiar with the safe preparation, use, storage, and disposal of reagents commonly encountered in inorganic chemistry laboratories
B4. To increase familiarity with and competence in the major synthetic techniques and methods of analysis available to the modern inorganic chemist
B5. To develop excellent laboratory note-taking skills and to appreciate the importance of the laboratory notebook as a legal document
B6. To understand the ethical obligation of experimentalists in reporting results of their experiments
B7. To communicate experimental design and interpretation of experimental results in the standard format encountered in the scientific literature
Adriana Dinescu  
CSC 408  
phone 408-4623  
email: adriana.dinescu@wilkes.edu

**Schedule for Fall ‘15**

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